Total Synthesis of (-)-Histrionicotoxin 285A and (-)-Perhydrohistrionicotoxin

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ABSTRACT



Starting from commercially available (S)-glycidol, and via a common intermediate, the total synthesis of (-)-histrionicotoxin 285A and (-)-perhydrohistrionicotoxin has been achieved. Key to this synthesis was the efficient construction of a six-membered, chiral, cyclic nitrone.

The histrionicotoxins (HTXs) are a family of spirocyclic piperidine alkaloids isolated in 1971 from the Colombian frog *Dendrobates histrionicus*.¹ Their intriguing molecular architecture coupled with potent inhibition of the nicotinic acetylcholine receptor² has made them attractive targets for total synthesis. Most effort has been directed to histrionic-otoxin 283A **1** and the unnatural perhydrohistrionicotoxin **2**.³ In this paper, we report the first synthesis of (–)-HTX

285A **3**, the most abundant⁴ and potent^{1,2b} alkaloid within this family, containing both a terminal *cis*-enyne and an allene side chain (Figure 1).

We sought a new enantioselective synthetic route, based upon the nitrone dipolar cycloaddition approach,⁵ which would necessitate the efficient construction of a six-membered chiral cyclic nitrone. Inspired by the work of Goti⁶ and Grigg⁷ involving the intramolecular *N*-alkylation of oximes, we reasoned that the required nitrone **15** could be prepared by ring closure of the advanced oxime precursor **14**.

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Commercially available *S*-glycidol **4** was transformed into the chiral lactone **9** in five steps (65% overall yield) following the general strategy described by Forsyth⁸ (Scheme 1). The lactone **9** was efficiently opened⁹ in excellent yield with the lithium acetylide derivative of **10** to give the alkynone **11**, which was catalytically hydrogenated to afford the fully saturated derivative **12**. The keto-alcohol **12** showed no evidence of closure to the corresponding cyclic hemiketal tautomer.

Mesylation of the alcohol **12** gave the mesylate **13**. Treatment of the keto-mesylate **13** with NH₂OH gave the presumed oxime **14** which showed no tendency to cyclize at room temperature. However, following the observation of Py,¹⁰ it was found that simply heating the mesylate **13** in the presence of an excess of NH₂OH in EtOH at 70 °C achieved both the conversion of **13** into the oxime **14** and subsequent intramolecular *N*-alkylation (with inversion of configuration) to give the nitrone **15**. Without purification, the unstable nitrone **15** was immediately protected as its styrene adduct **16**. In this manner, the mesylate **13** was efficiently converted into isoxazolidine **16** in an overall yield of 62%.

Elaboration of the acetal **16** to the *Z*- α , β -unsaturated nitrile **17** was achieved by mild transacetalization of **16** to liberate the aldehyde, followed by olefination using the superior, modified Peterson reagent described by Kojima.¹¹ This latter reaction gave in excess of 95:5 *Z/E* selectivity (trace amounts of the *E*-isomer could be chromatographically removed). The nitrile **17** was obtained in 81% yield from acetal **16**. The nitrile **17** was subjected to a microwave (MW) thermally induced 1,3-dipolar cycloreversion (with accompanying extrusion of styrene) followed by an intramolecular cycloaddition process under thermodynamic control to furnish the required (6,6,5)-tricycle **18**.⁵

Thus, the absolute configuration of **18** is determined by *S*-glycidol **4**; the remaining three stereocenters are induced by a diastereoselective dipolar cycloaddition to the *Z*- α , β -unsaturated nitrile. This approach produces the core HTX precursor **18** in 13 steps and 19% overall yield from *S*-glycidol **4**.





With tricycle **18** in hand, we first explored the synthesis of (–)-perhydrohistrionicotoxin **2** (Scheme 2). The silyl protecting group on **18** was removed with AcOH-buffered TBAF¹² (without AcOH slow epimerization of the axial nitrile moiety was observed) to give alcohol **19**. Alcohol **19** was activated as its mesylate and displaced with cyanide to provide the dinitrile **20**. The dinitrile **20** was converted into the dialdehyde followed by Wittig olefination to give exclusively the *Z*,*Z*-dialkene **21**.¹³ Owing to the sluggish reactivity of the *N*–*O* bond within **21** toward hydrogenolysis, a direct, one-step reduction to perhydrohistrionicotoxin **2** was abandoned. Instead, the isoxazolidine moiety was selectively reduced under Brandi (SmI₂) conditions,¹⁴ and advantageously, gave rise to a new (albeit unnatural) member of

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Scheme 2. Synthesis of (-)-Perhydrohistrionicotoxin 2



the histrionicotoxin family, namely, **22**. Catalytic hydrogenation of dialkene **22** afforded (–)-perhydrohistrionicotoxin **2**. The spectroscopic data including the specific rotation of the hydrochloride salt of (–)-perhydrohistrionicotoxin **2** matched that previously reported.^{15,16}

We then investigated the more synthetically challenging histrionicotoxin 285A **3** (Scheme 3). The nitrile **18** was converted into the aldehyde and then olefinated under Stork–Wittig conditions¹⁷ to afford Z-vinyl iodide **23**. The silyl-protecting group was cleaved under acidic conditions¹⁸ to provide the alcohol **24**which underwent Sonogashira coupling¹⁹ to give Z-enyne **25**. We anticipated introducing the terminal allene moiety by iodide displacement with allenyl lithium. Mesylate activation of the alcohol **25** and cyanide displacement gave the nitrile **26**. A two-step reduction of **26** yielded the primary alcohol **27** which was transformed to the alkyl iodide **28** by mesylation followed by displacement with iodide. The key displacement reaction of iodide **28** with an excess of allenyl lithium in the presence of HMPA was remarkably clean and conScheme 3. Synthesis of (-)-Histrionicotoxin 285A 3



comitantly cleaved the triisopropylsilyl protecting group to afford the allene derivative **29** in excellent yield. Finally, SmI₂ cleavage of the N–O bond afforded (–)histrionicotoxin 285A **3**. The spectroscopic data matched that previously reported.^{1,16a,20}

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Supporting Information Available: Experimental procedures and compound characterization for **2**, **3**, **5**, **7**–**9**, **11**–**13**, and **16**–**29**. This material is available free of charge via the Internet at http://pubs.acs.org.

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